

Year 9 Curriculum Overview Subjects and qualifications: GCSE Separate Sciences – Biology, Chemistry and Physics

GCSE Biology, Chemistry and Physics:

The Big Idea Principle.

Our KS4 Science curriculum at ICC builds on our KS3 approach, using AQA's big ideas principle so that the generalisations, principles and models which connect Scientific concepts are at the heart of student learning. We believe this is how students learn to see the world analytically, to explain phenomena and make predictions – all skills they need for their next stage of scientific learning. Content across KS4 is colour coded and mapped against the big idea principles from our key stage 3 curriculum so that students can clearly recognise the progression in their learning and as a result are more confident in applying the transferable skills that they have developed in their prior learning to new and unfamiliar ideas in their GCSE studies.

The Big Idea's studied in Biology lessons: Organisms and Ecosystem

Earth	Ecosystem	Organisms	Reactions	Forces	Waves	Matter	Genes	Electromagnets	Energy
The Big Idea's studied in Chemistry lessons: Matter and Reactions									
Earth	Ecosystem	Organisms	Reactions	Forces	Waves	Matter	Genes	Electromagnets	Energy
	's studied in	Physics loss	ons: Enorgy	Electromagn	ots and Matte				
	s studied in	Filysics less	<u>ons.</u> Energy,	Electromagn					
Earth	Ecosystem	Organisms	Reactions	Forces	Waves	Matter	Genes	Electromagnets	Energy



GCSE		Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2
Bio	logy						
	Unit(s):	Cell Biology	Cell Biology and Bioenergetics	Bioenergetics	Organsiation	Organsiation	Organisation
Year 9	Key Skills:	HSW: Analysing patterns, drawing conclusions, collecting data, comparing equipment, using apparatus, biological drawing Maths: Calculations, rearranging equations, standard form, orders of magnitude, standard units and conversions Literacy: Labelling diagrams, discussion, debate, long answer questions	HSW: Using apparatus, collecting data, presenting data, interpreting data, applying practical to real life situations, draw conclusions Maths: Balancing equations, drawing graphs, calculating light intensity (HT Only), calculating rate of change Literacy: Labelling diagrams, discussion, debate, long answer questions, using word	HSW: Using apparatus, collecting data, presenting data, interpreting data, applying practical to real life situations, draw conclusions Maths: Rearranging equations, standard form, balancing equations, rate of change Literacy: Labelling diagrams, discussion, debate, oracy, long answer questions, using	HSW: Using models, discuss limitations, draw conclusions, collect data, plan variables, test hypothesis Maths: Graphs, calculating percentages, using the mean, median and mode, significant figures, analysing correlation, make estimations of results, decimals Literacy: Using qualitative data, labelling diagrams, defining keywords, discussion, long answer questions	HSW: Evaluating models, discuss limitations, presenting data, interpreting data, using and selecting apparatus, Maths: Graphs, calculating percentages, mean, median and mode, significant figures, analysing correlation, make estimations of results, decimals Literacy: Using qualitative data, labelling diagrams, defining keywords, debate, long answer questions	HSW: Discuss limitations, draw conclusions, collect data, plan variables, test hypothesis, presenting data, interpreting data, interpreting data, Maths: Graphs, calculating percentages, mean, median and mode, significant figures, analysing correlation, make estimations of results, decimals Literacy: Using qualitative data, labelling diagrams, defining keywords, discussion, debate_long
		using articles	equations, justification	articles, using word equations,	justification, comprehension	justification, comprehension	answer questions, justification,



Assessment:	Extended	Extended	Extended	Extended	Extended	Extended
	Response:	Response:	Response	Response: N/A	Response	Response
	1. Explaining	1. Evaluating	1. Explaining how	•	1. Explaining how	1. Use the
	movement of	stem cells	to investigate	Cumulative	the villi are	information from a
	particles	2. Comparing the	the effect of light	assessment:	adapted to	table and explain
	between cells.	effectiveness of	intensity on the	Cell Biology	maximise the	why beta blockers
		different	rate of	Bioenergetics	rate of	change breathing
	Practical:	antibiotics at	photosynthesis.		absorption.	rate during exercise.
	RP1 Microscopes:	killing different	2. Use	Practical:	2. Explain how	
	Use a light	bacteria.	information	RP3 Food tests:	starch in a	End of unit test:
	microscope to	3. Describe and	from graphs to	Use qualitative	vegetable is	Organisation
	observe, draw and	Explain factors	explain	reagents to test for a	produced.	
	label a selection of	that affect the	changes in	range of		
	plant and animal	rate of	neart rate during	carbohydrates, lipids	Cumulative	Practical: N/A
	cells. Magnification	pnotosyntnesis.	exercise.	and proteins. To	assessment:	
	scale must be	End of unit toot	End of unit toot	include: Benedict's	Cell Biology	
	Included.	Coll Rielegy	Piconorgotion	test for sugars;	Bioenergetics	Homework [.]
	Investigate the	Cell Blology	Dibenergetics	iodine test for	Organisation	Organisation
	effect of a range of	Practical: N/A	Brootical	starch; and Biuret		retrieval questions
	concentrations of		Practical:	reagent for protein.	Practical:	
	salt or sugar		RP3 Destacymthesis		RP4 Enzymes:	
	solutions on the	Homowork	Investigate the	Homework:	Investigate the	
	mass of plant	Piconerrotice	effect of light	Organisation MCQ,	effect of pH on the	
	tissue.	Sonoca Coll	intensity on the rate			
		Biology retrieval	of photosynthesis	retrieval questions	amylase enzyme.	
	Homework: Cell	auestions	using an aquatic		Homovionly	
	Biology MCQ and	questions	organism such as		Homework:	
	Seneca, Organisms		pondweed.		Organisation	
	retrieval questions				Biognargatics	
			Homework:		retrieval questions	
			Bioenergetics		Tellieval questions	
			MCQ, Cell Biology			
			retrieval questions			
Retrieval:	Living organisms,	Health and	Living organisms,	Animal and Plant	Aerobic and	Living organisms,
	cells, inheritance	disease, transport	nutrients within	cells, transport of	anaerobic	Animal and Plant
	of genes	of substances in	food, diffusion,	substances in	respiration,	cells, transport of
	-	cells, function of	osmosis and	cells	photosynthesis	substances in
		mitochondria and	active transport			cells, inheritance
		chloroplasts				of genes



GCSE Chomistry		Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2
Cinc	Unit(s):	Atomic structure and the periodic table	Bonding structure and the properties of matter	Bonding, structure and the properties of matter and Quantitative chemistry 1	Energy Changes	Chemical Changes	Chemical Changes
	Key Skills:	 HSW: Predicting reactions and properties, using models, separating mixtures Maths: Balancing equations, percentages, fractions, SI units and orders of magnitude Literacy: long answer questions, historical timelines 	 HSW: Predicting HSW: Models and analogies, model drawings Maths: 3D models, Equations Literacy: Long answers questions, researching uses of substances, using articles 	HSW: Models and analogies, model drawings Maths: 3D models, Equations, Percentages, writing and balancing equations, chemical measurements, uncertainty Literacy: Long answer questions, using articles, researching uses of substances, Writing equations	HSW: Temperature changes experiment, drawing energy profiles Maths: Thermometer readings, calculating means, measuring differences Literacy: Using chemistry in everyday life work	HSW: Chemical reactions practical's, electrolysis, making salts Maths: pH scales and orders of magnitude, concentrations, equations Literacy: Writing chemical equations, deducing products	HSW: Chemical reactions practical's, electrolysis, making salts Maths: pH scales and orders of magnitude, concentrations, equations Literacy: Writing chemical equations, deducing products



Assessment:	Extended	Extended	Extended	Extended	Extended	Extended
	Response:	Response:	Response:	Response:	Response:	Response:
	1. Describe the	1. Use information	1. Compare ionic	1. Explain reaction	1. Plan an	2. Describe how to
	practical	to explain	and covalent	profiles	experiment to	make pure
	techniques for	conclusions for	bonding		test reactivity of	crystals
	separating	metal oxides.	2. Calculate	End of unit test:	metals	
	mixtures.	2. Describe what	relative formula	Energy changes		End of unit test:
	2. Explaining the	happens	mass and		Cumulative	Chemical Changes
	plum pudding	magnesium and	explain the law	Practical: RQP 10	assessment:	
	model	iodine react and	of conservation	endothermic and	Bonding,	Practical: RQP 8
		3. Explain why a	of mass	exothermic	structure, and the	soluble salts:
	End of unit test:	nign temperature		reactions:	properties of	Preparation of a
	Atomic structure	is needed to melt	End of unit test:	Investigate the	matter	pure, dry sample of a
	and the periodic	inagnesium	Bonding,	variables that affect	Quantitative	soluble salt from an
	table	iodide.	structure, and the	temperature changes	chemistry 1	insoluble oxide
	Dreatical N/A	Cumulativa	properties of	in	Energy changes	
	Practical: N/A	Cumulative	matter	reacting solutions.	Chemical changes	Homework:
		assessment:				Chemical changes
	Homework:	Atomic structure	Practical: N/A	Homework:	11	Seneca and MCQ /
	Atomic structure	and the periodic		Energy changes	Homework:	Quantitative
	and the periodic	structure, and the	Homework:	MCQ and Seneca /	Chemical changes	chemistry 1 and
	table Seneca /	properties of matter	Quantitative	Quantitative	Seneca,	Energy changes
		properties of matter	chemistry 1	chemistry 1 retrieval	Quantitative	retrieval questions
	questions	Homework	Seneca / Bonding,	questions	Energy changes	
		Bonding structure	structure, and the		retrieval questions	
		and the properties	properties of		remeval questions	
		of matter Seneca	matter retrieval			
		and MCQ / Atomic	questions			
		structure and The				
		periodic table				
		retrieval questions				
Retrieval:	Separating	Periodic table,	Atomic structure,	Atomic structure,	Chemical	Metals and non-
	mixtures, atomic	elements,	drawing ions,	bonding, drawing	equations, neutral	metals, pH,
	structure	reactions and	properties of	ions, properties of	substances,	periodic table,
		equations	metals and non-	metals and non-	elements vs	bonding, chemical
			metals. Elements	metals. Chemical	compounds.	equations.
			electronic	reactions and	states of matter	reactions
			configuration	equations element		experimental
				and periodic table		procedure



GCSE		Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2	
Physi	ics							
	Unit(s):	Energy	Energy	Electricity	Electricity	Particle Model	Particle Model	
Year 9	Key Skills:	HSW: Development of scientific thinking; Analysis and evaluation; Scientific vocabulary; quantities, units, symbols, and nomenclature Maths: Arithmetic and numerical computation; Handling data; Algebra Literacy: Oracy via discussion; Written equations	HSW: Development of scientific thinking; Scientific vocabulary, quantities, units, symbols, and nomenclature Maths: Arithmetic and numerical computation; Algebra; Graphs Literacy: Evaluating; Understanding key vocabulary; Written equations	HSW: Development of scientific thinking; Analysis and evaluation Maths: Arithmetic and numerical computation; Algebra; Graphs Literacy: Written equations; Sequencing information	HSW: Development of scientific thinking; Scientific vocabulary, quantities, units, symbols, and nomenclature Maths: Arithmetic and numerical computation; Algebra; Graphs Literacy: SECC; Dialogic lesson	HSW: Development of scientific thinking; Scientific vocabulary, quantities, units, symbols, and nomenclature Maths: Arithmetic and numerical computation; Algebra; Graphs Literacy: Oracy via discussion; Written equations	HSW: Development of scientific thinking; Scientific vocabulary, quantities, units, symbols, and nomenclature Maths: Arithmetic and numerical computation; Algebra; Graphs Literacy: Understanding key vocabulary; Evaluating	



Assessment:	Extended Response: 1. Stating advantages and disadvantages s of the two methods of generating electrical energy. Practical: RQP 14 Specific Heat Capacity: An investigation to determine the specific heat capacity of one or more materials. Homework: Energy Seneca	Extended Response: N/A End of unit test: Energy Homework: Energy Seneca / Energy MCQ	Extended Response: 1. Describe how to determine resistance from simple circuit Cumulative assessment: Energy Electricity Practical: RQP 15 Resistance: Use circuit diagrams to set up and check appropriate circuits to investigate the factors affecting the resistance of electrical circuits.	Extended Response: N/A End of unit test: Electricity Practical: RQP 16 I-V characteristics: Use circuit diagrams to construct appropriate circuits to investigate the I–V characteristics of variety of circuit elements including a filament lamp, a diode and a resistor at constant Temperature. Homework: Electricity Seneca / Energy, Electricity MCQ	Extended Response: 1. Compare properties of solids and gases Cumulative assessment: Energy Electricity Particle Model Practical: RQP 17 Density: Use appropriate apparatus to make and record the measurements needed to determine the densities of regular and irregular solid objects and liquids Homework: Particle Model and	Extended Response: 1. Describe method for determining density of irregular shape End of unit test: Particle Model Practical: N/A Homework: Energy, Electricity, Particle Model MCQ
			Homework: Electricity Seneca / Energy MCQ		Particle Model and Atomic Structure Seneca / Energy, Electricity and Particle Model MCQ	
Retrieval:	KS2 and 3 Energy	Energy and KS3 Electricity	Energy and Electricity	Energy and Electricity	Energy, Electricity and KS3 Particle Model	Energy, Electricity and Particle Model